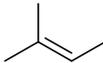
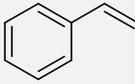
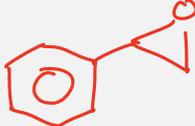
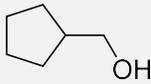
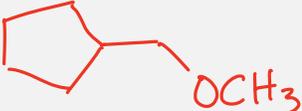
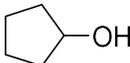
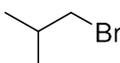
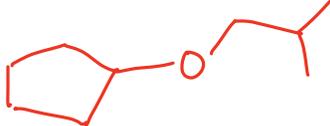
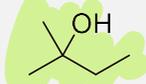
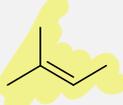
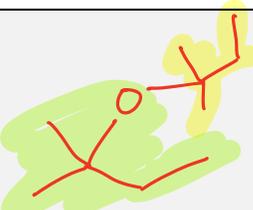


**Chapter 13 Worksheet – Ethers & Epoxides****13A. Alkene Reactions** - Draw the product of the reaction between each alkene (1-3) with reagents (a)-(c).

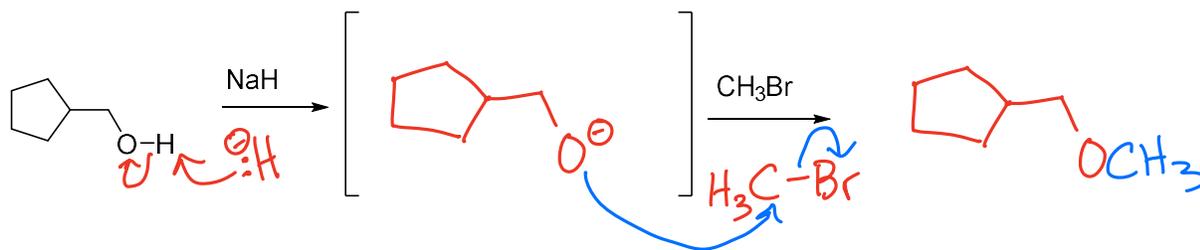
	Starting Material	Reagents & translation	Draw the Product
1		(a) 1. $\text{Hg}(\text{OAc})_2, \text{H}_2\text{O}$ 2. $\text{NaBH}_4$  Oxymercuration / demercuration <i>Mercury (II) acetate and water,</i> <i>followed by sodium borohydride</i>	
2		(b) 1. $\text{Hg}(\text{OAc})_2, \text{CH}_3\text{OH}$ 2. $\text{NaBH}_4$  Alkoxymercuration / demercuration <i>Mercury (II) acetate and alcohol,</i> <i>followed by sodium borohydride</i>	
3		(c) <i>m</i> -CPBA  <i>meta-chloroperoxybenzoic acid</i>	

**13B. Alcohol Reactions** - Draw the product between each alcohol (4-6) and reagents (d – f).

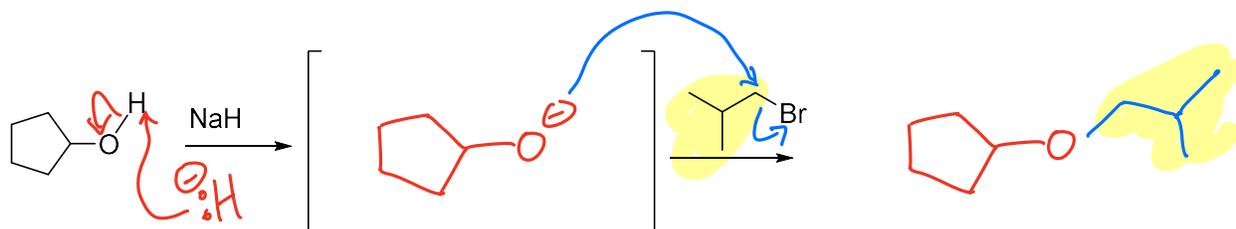
	Starting Material	Reagents & translation *know this mechanism	Draw the Product
4		*(d) 1. $\text{NaH}$ 2. $\text{CH}_3\text{Br}$  <i>Sodium hydride followed by methyl bromide</i>	
5		*(e) 1. $\text{NaH}$ 2.   <i>Sodium hydride followed by isobutyl bromide</i>	
6		(f) 1. $\text{Hg}(\text{OAc})_2$  2. $\text{NaBH}_4$  Alkoxymercuration / demercuration <i>Mercury (II) acetate and alkene,</i> <i>followed by sodium borohydride</i>	

**13B Alcohol Reaction Mechanisms** – Draw the reaction mechanism and product for the reactions below.

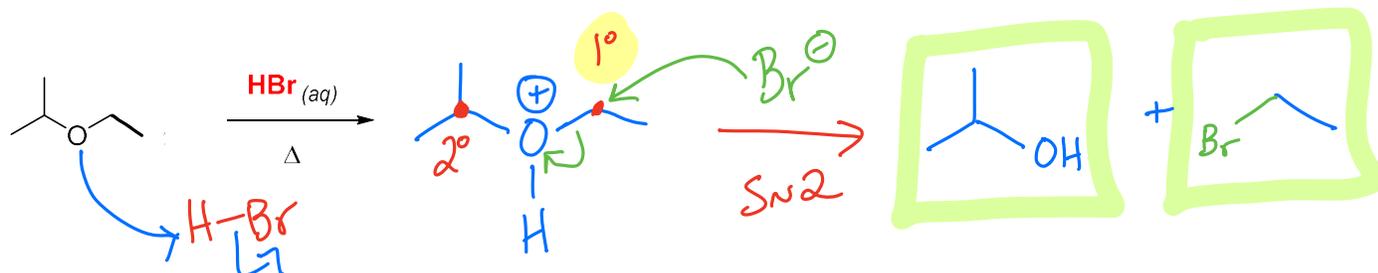
(4d)



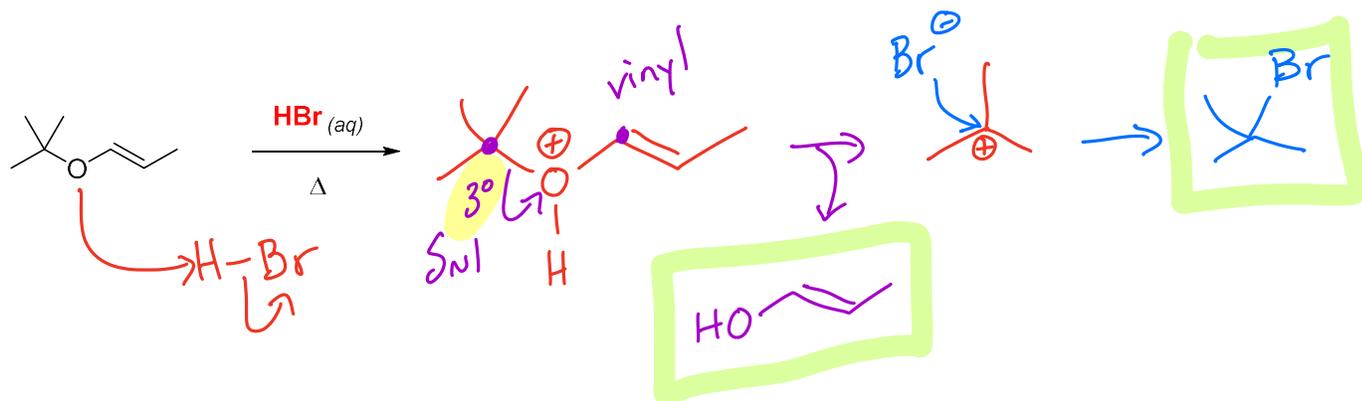
(5e)


**13C Acidic Ether Cleavage.** React ethers 7 & 8 with aqueous HBr and heat. Draw the mechanism for each.

7.

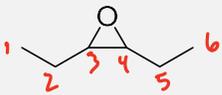
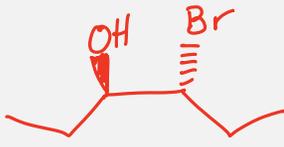
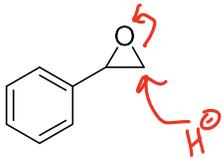
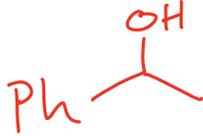
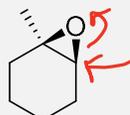
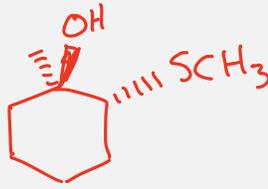
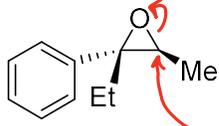
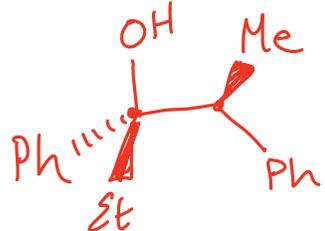
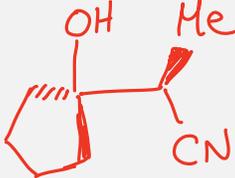
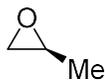
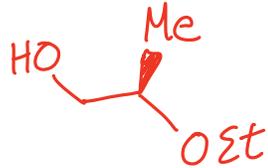


8.



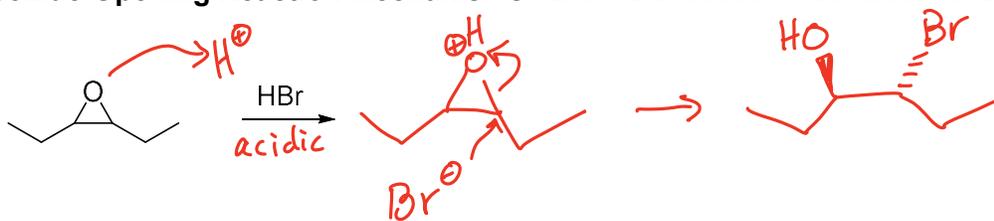
**13D Epoxide-Opening Reactions.** Draw the product of the reaction between epoxides **9-14** and reagents **g-l**.

- Consider drawing the mechanisms on the next page to come up with the product.

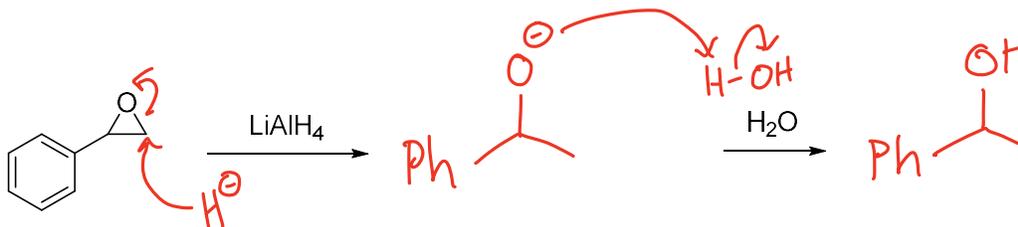
	<b>Starting Material</b>	<b>Reagents &amp; translation</b> *know this mechanism	<b>Draw the Product</b> with stereochemistry (wedges & dashes on chiral centers)
9		<p><b>*(g) HBr</b> <i>acidic</i> Hydrobromic acid</p>	
10		<p><b>*(h) 1. LiAlH<sub>4</sub></b> <i>basic</i> 2. H<sub>2</sub>O <i>lithium aluminum hydride followed by water</i></p>	
11		<p><b>*(i) 1. NaSCH<sub>3</sub></b> <i>basic</i> 2. H<sub>2</sub>O <i>Sodium methyl thiolate followed by water</i></p>	
12		<p><b>*(j) 1. MgBr</b> <i>basic</i> 2. H<sub>2</sub>O <i>Phenyl magnesium bromide followed by water</i></p>	
13		<p><b>*(k) 1. NaCN</b> <i>basic</i> 2. H<sub>2</sub>O <i>Sodium cyanide followed by water</i></p>	
14		<p><b>*(l) EtOH, [H<sub>2</sub>SO<sub>4</sub>]</b> <i>Ethanol with sulfuric acid catalyst</i></p>	

**13D Epoxide-Opening Reaction Mechanisms - Draw the reaction mechanism and product for each reaction.**

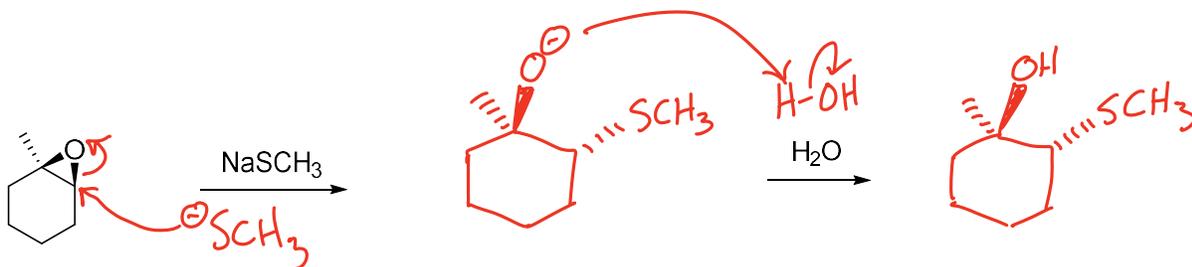
(9g)



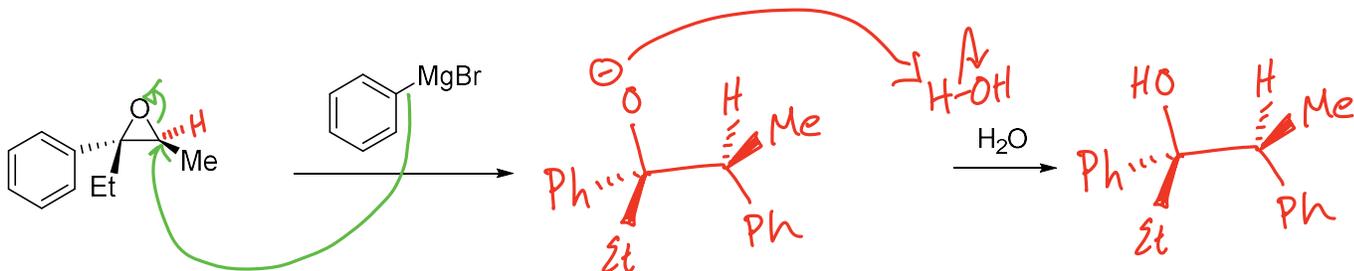
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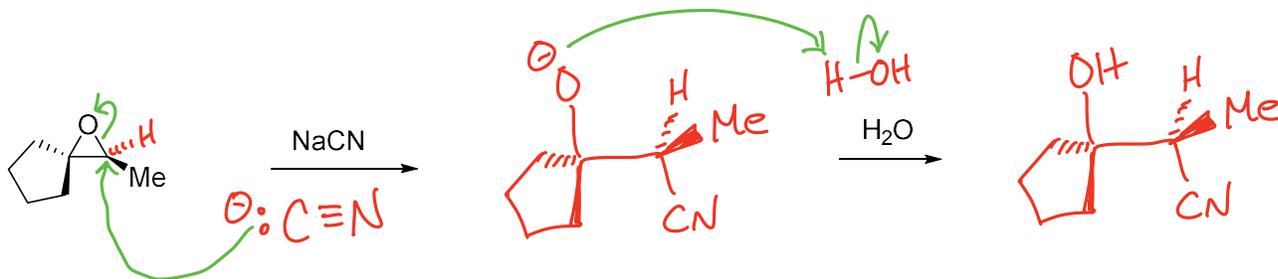
(11i)



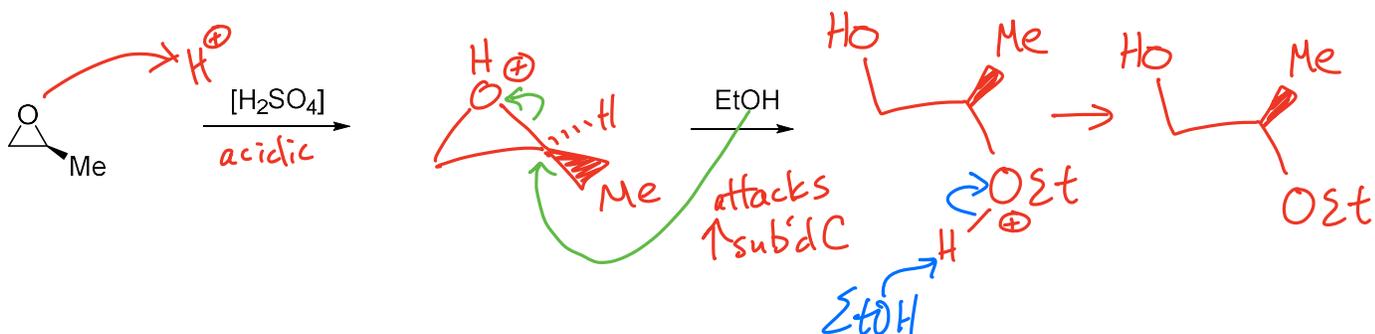
(12j)



(13k)

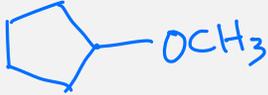
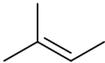
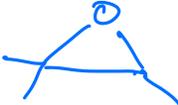
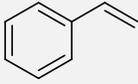
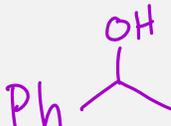
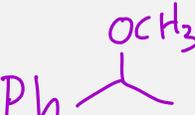


(14l)

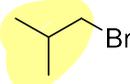
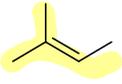
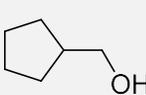
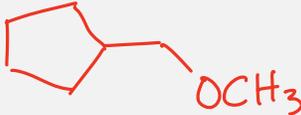
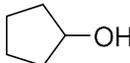
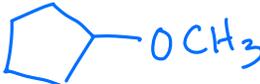
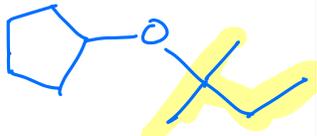
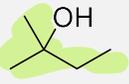
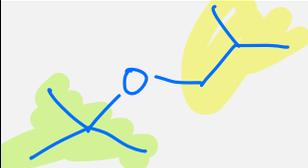


## BONUS: Mix &amp; Match with Reaction Bootcamp!

## 13A Alkene Reactions

Starting Material		(a) 1. Hg(OAc) <sub>2</sub> , H <sub>2</sub> O 2. NaBH <sub>4</sub>	(b) 1. Hg(OAc) <sub>2</sub> , CH <sub>3</sub> OH 2. NaBH <sub>4</sub>	(c) <i>m</i> -CPBA
1				
2				
3				

## 13B Alcohol Reactions

Starting Material		(d) 1. NaH 2. CH <sub>3</sub> Br	(e) 1. NaH 2. 	(f) 1. Hg(OAc) <sub>2</sub>  2. NaBH <sub>4</sub>
4				
5				
6				

## BONUS: Mix &amp; Match with Reaction Bootcamp!

Reagents	9.	11.	14.
<p><b>(g) HBr</b> Hydrobromic acid</p> <p><i>S<sub>N</sub>1 acidic</i> Br goes to more sub'd C</p>			
<p><b>(h) 1. LiAlH<sub>4</sub></b> 2. H<sub>2</sub>O</p> <p>lithium aluminum hydride followed by water</p> <p><i>basic S<sub>N</sub>2</i> H goes to less sub'd C</p>			
<p><b>(i) 1. NaSCH<sub>3</sub></b> 2. H<sub>2</sub>O</p> <p>Sodium methyl thiolate followed by water</p> <p><i>basic S<sub>N</sub>2</i> SCH<sub>3</sub> goes to less sub'd C</p>			
<p><b>(j) 1. PhMgBr</b> 2. H<sub>2</sub>O</p> <p>Phenyl magnesium bromide followed by water</p> <p><i>basic S<sub>N</sub>2</i></p>			
<p><b>(k) 1. NaCN</b> 2. H<sub>2</sub>O</p> <p>Sodium cyanide followed by water</p> <p><i>basic S<sub>N</sub>2</i></p>			
<p><b>(l) EtOH, [H<sub>2</sub>SO<sub>4</sub>]</b> Ethanol with sulfuric acid catalyst</p> <p><i>S<sub>N</sub>1 acidic</i></p>			

*EtO goes to more sub'd C, if there is one*