<u>Chapter 20 Worksheet – Carboxylic Acids & Friends</u>

20A. HYDROGEN & CARBON NUCLEOPHILES.

Draw the product of each reaction: starting material + reagent \rightarrow Product.

Starting Material		Reagents & translation *know this mechanism	Draw the Product Pay attention to the amount of reagent added!
1	ОНО	*(a) xs NaBH ₄ , MeOH sodium borohydride in methanol	r dy ditorition to the dimodrit of reagonit added.
2	OCI	* (b) 1. xs LiAlH₄ 2. H₂O lithium aluminum hydride followed by water	
3		* (c) 1. xs CH ₃ MgBr 2. H ₂ O Ethyl magnesium bromide followed by water	
4	∕ ∕ ∕ CN	* (d) 1. PhMgBr (1 mol) 2. H ₃ O ⁺ Phenyl magnesium bromide followed by aqueous acid	
5	O C	* (e) Ph ₂ CuLi (1 mol) Gilman reagent: diphenyl organocuprate	
6	NC	(b) 1. xs LiAlH₄ 2. H₂O lithium aluminum hydride followed by water	
7	O NMe ₂	(b) 1. xs LiAlH₄ 2. H₂O lithium aluminum hydride followed by water	

20A. Mechanisms – Acid Derivatives with hydrogen- and carbon-nucleophiles.

- Draw the <u>arrow-pushing mechanism</u> for each reaction, including all charged <u>intermediates and product</u>.

(3b) Acid anhydride reduction

$$\begin{array}{c|c}
0 & 0 \\
\hline
2. H_2O
\end{array}$$

(5c) Grignard addition to acid chloride

$$\begin{array}{c} O \\ CI \end{array} \qquad \begin{array}{c} 1. \text{ xs CH}_3\text{MgBr} \\ \hline \\ 2. \text{ H}_2\text{O} \end{array}$$

(4d) Grignard addition to nitrile

(2e) Gilman addition to acid chloride

20B. Nucleophilic Acyl Substitution – Mix & Match with Reaction Bootcamp! – required, NOT bonus

• Draw the product of each reaction: starting material + reagent → Product.

React each friend with each reagent and draw the product in the box		*(f) O Ph OH pyridine	*(g) H₃O ⁺	* (h) OH H ⁺ (acid catalyst)	*(i) xs NH₃
2	o c				
3		No Reaction			No Reaction
8	ОН	No Reaction	No Reaction		No Reaction
9	H ₃ CO H ₃ CO	No Reaction			
7	O NMe ₂	No Reaction			No Reaction
4	∕ ∕ CN	No Reaction		No Reaction	No Reaction

20B. Nucleophilic Acyl Substitution Mechanisms

- Draw the <u>arrow-pushing mechanism</u> for each reaction, including all charged <u>intermediates and product</u>.

(2i) Aminolysis of acid chloride

(3g) Hydrolysis of acid anhydride

(8h) Fischer esterification - alcohololysis of carboxylic acid

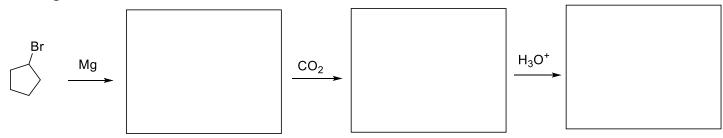
$$\bigcirc O \qquad \bigcirc O$$

(4g) Hydrolysis of nitrile

20C. Miscellaneous Reactions that didn't fit nicely into tables

- Fill in the box: Draw the <u>product of each reaction</u>: **starting material + reagent → Product**.
- Mechanisms are helpful, but are not required here (probably not enough space anyway).

10. Grignard formation and addition to carbon dioxide



11. Addition of thionyl chloride to carboxylic acid

12. Addition of thionyl chloride to amide – corrected / updated on 2/18

13. Partial reduction of an ester with DIBAH (diisobutyl aluminum hydride)

$$OCH_3 \qquad 1. DIBAH \\ \hline 2. H_2O$$

BONUS – optional, extra credit

- 1. Make up a molecule that includes a carboxylic acid and all of its friends!
 - a. Acid chloride
 - b. Acid anhydride
 - c. Carboxylic acid
 - d. Ester
 - e. Amide
 - f. Nitrile
- 2. Add decorations to make your molecule look like an animal, creature, or something else that's fun