

Name \_\_\_\_\_ Partner Name \_\_\_\_\_

TA Name \_\_\_\_\_ Section Letter \_\_\_\_\_ Day \_\_\_\_\_ Time \_\_\_\_\_

## Experiment 2 Worksheet – Perkin Condensation: Synthesis of *trans*-Cinnamic Acid

Use as reference for notebook preparation – submit on Canvas this individually after lab

### Pre-Lab Requirements

1. **Dress for lab** – see safety rules – arrive a few minutes early
2. Copy these templates into your lab notebook – contact instructors for alternate accommodations
  - Fill in the **purpose with structures and reagent table**
  - **Procedure Diagrams** – must be complete before you can start the lab

### A. Experimental Purpose and Perkin Condensation Reaction Scheme

### B. Reagent Table

Refer to the procedure for amounts and safety table for hazards; find the chemical properties on Wikipedia!

Name	Volume	Density	Mass	MW	mmol	Equiv*	Boiling or melting point	Hazards
Benzaldehyde (PhCHO)								
Acetic anhydride (Ac <sub>2</sub> O)								
Potassium acetate	-	-						
<i>trans</i> -cinnamic acid (product)	-							

\* **Equiv** = molar equivalents of reaction components with respect to the limiting reagent (benzaldehyde)

- Acetic Anhydride (reagent): divide the mmol of reagent by the mmol of salicylic acid

**C. Procedure Diagrams** - on as many pages as needed.

- **All labeled equipment, chemical names with amounts, transfers, cleanup & safety notes**
  - Help w diagrams: Slugs@home Exp 2 website & class notes

Experiment segments

1. **Reaction setup** – all equipment and chemicals (name, structure, and amount)
2. **Reaction workup** – flow chart / diagrams of separatory funnel contents and all solution transfers
3. **Analysis** – NMR and IR sample preparation; sketches of spectra, identifying key signals

## E. Data & Observations

Volume of benzaldehyde \_\_\_\_\_ mL

Theoretical yield of t-cinnamic acid \_\_\_\_\_ mg

*Theoretical Yield Calculation:*

*Product Loss– list the amount and specific reason for the loss*

Crude yield of product, *trans*-cinnamic acid = \_\_\_\_\_ mg

**III. Spectroscopic Analysis** – Analyze the IR obtained in lab and NMR spectrum on the Slugs@home website. Identify any signals within the expected range. It is acceptable for a signal to be “not observed.”

**Benzaldehyde IR**

Functional Group	Bond	Expected Wavenumber Range (cm <sup>-1</sup> )	Observed Wavenumber (cm <sup>-1</sup> )

<sup>1</sup>H NMR of benzaldehyde  
(draw structure with labels)

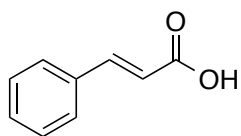
<sup>13</sup>C NMR of benzaldehyde

Signal	Integration (#H's)	Splitting	Chemical Shift Expected (ppm)	Chemical Shift Observed (ppm)	Signal	Chem Shift Expected (ppm)	Chem Shift Observed (ppm)
A					A'		
B					B'		
C					C'		
D					D'		
					E'		

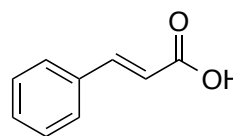
**III. Spectroscopic Analysis** - Analyze the IR obtained in lab and preview the NMR spectra on the Slugs@home website. Your NMR spectra will be shared on Canvas a few days after lab. Identify any signals within the expected range. It is acceptable for a signal to be "not observed."

***trans*-cinnamic acid IR**

Functional Group	Bond	Expected Wavenumber Range (cm <sup>-1</sup> )	Observed Wavenumber (cm <sup>-1</sup> )



<sup>1</sup>H NMR of *trans*-Cinnamic Acid



<sup>13</sup>C NMR

Signal	Integration (#H's)	Splitting	Chemical Shift Expected (ppm)	Chemical Shift Observed (ppm)	Signal	Chem Shift Expected (ppm)	Chem Shift Observed (ppm)
A					A'		
B					B'		
C					C'		
D					D'		
E					E'		
F					F'		
					G'		